

6-Aza-, 6-Oxa- and 6-Thia-4,5,6,7-Tetrahydrobenzimidazoles  
by Electrophilic Intramolecular Ring Closure of Imidazoles

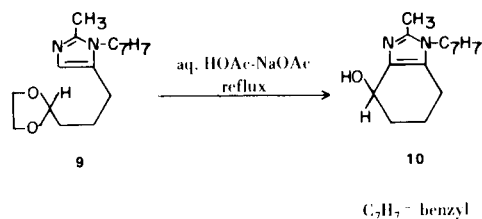
Hubert J. J. Loozen\*, Ben J. van der Beek, E. F. Godefroi and H. M. Buck

Department of Organic Chemistry, Eindhoven University of Technology, Eindhoven, The Netherlands

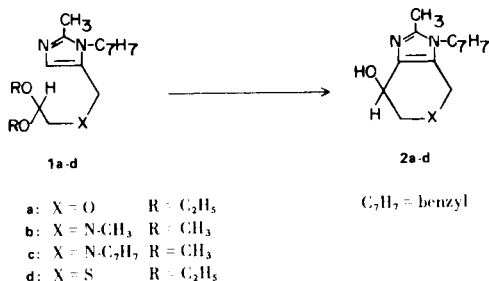
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It is well known that imidazoles are quite unreactive towards electrophilic substitution under the common strongly acidic conditions usually required (1a,b). However, under mildly acidic conditions (aqueous sodium acetate-acetic acid buffer), a facile hydroxymethylation of 1,2-disubstituted imidazoles with formaldehyde could be achieved, leading to the corresponding 5-hydroxymethyl derivatives (2).

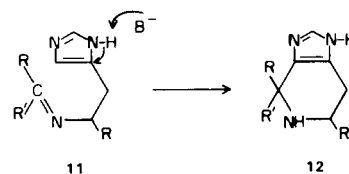
Recent work showed that under similar conditions benzimidazoles could be synthesized from imidazoles by intramolecular ring closure. An example was presented of the formation of a tetrahydrobenzimidazole (10) from the open-chain precursor (9) (3). In order to demonstrate the



generality of this reaction, we now wish to report the synthesis of 6-oxa-, 6-thia- and 6-azatetrahydrobenzimidazoles (2a-d) by means of intramolecular electrophilic ring closure of functionalized imidazoles (1a-d).



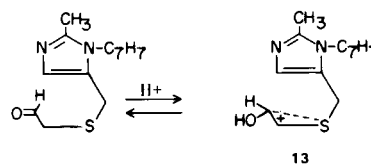
At present only some 6-aza-4,5,6,7-tetrahydrobenzimidazoles (12) are known and have been obtained by base catalyzed cyclization of Schiff bases (11) derived from histamine and histidine (4a-e). Essential in this approach was the absence of any substituent at the imidazole nitro-

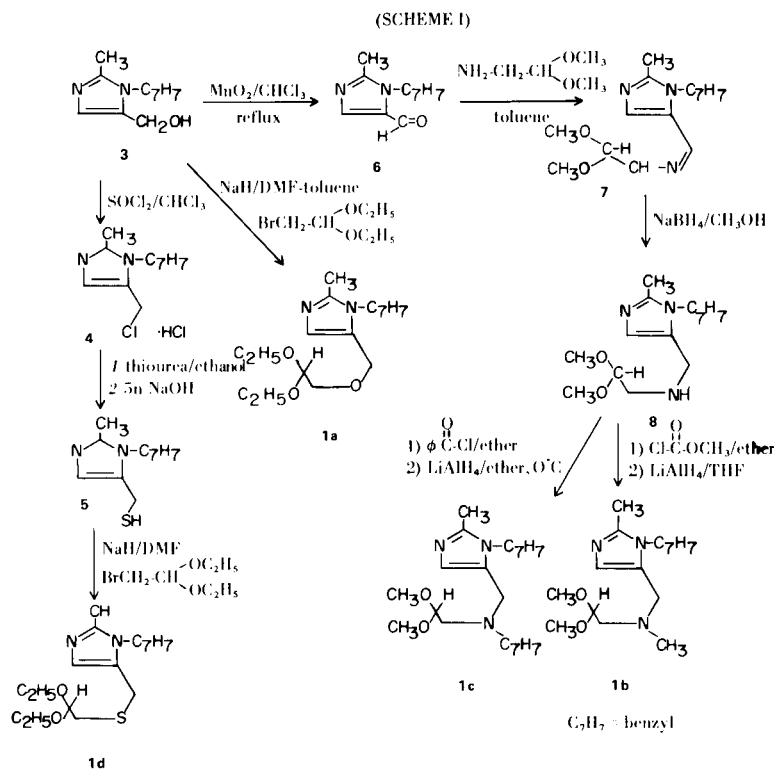


The preparation of the required substrates for cyclization (1a-d) is outlined in scheme 1. The cyclizations were monitored with thin layer chromatography. On refluxing 1a in an aqueous solution containing acetate buffer (pH ~ 5) after about 7 hours starting material had disappeared. Together with a considerable amount of tarry material two products were formed. Work-up and chromatography showed one compound to be the desired cyclic ether 2a, obtained in 28% yield. The other product proved to be alcohol 3, which must undoubtedly be the result of ether cleavage.

Cyclization reactions of compounds 1b and 1c under identical conditions as for 1a were also completed after seven hours.

In both cases relatively large amounts of tarry material were formed. Chromatography of the reaction products afforded 23% of 2b from 1b and 21% of 2c from 1c. The structures could be easily confirmed with nmr, ir and elemental analyses. On reacting 1d in buffered solution reaction was completed after two hours and one single product had been formed. Chromatography over a short column afforded in 65% yield the cyclic thioether 2d. The high rate and relative cleanliness of this reaction might be attributed to anchimeric assistance of sulfur in stabilizing the intermediate carbonium ion in 13. To the best of our knowledge such a pronounced effect has hitherto never been reported for a similar electrophilic substitution reaction.





## EXPERIMENTAL

## General.

Melting points were determined on a Mettler apparatus and are uncorrected. Nmr data were obtained with a Varian T-60A spectrometer (solutions in deuteriochloroform). Infrared spectra were recorded on a Perkin-Elmer 237. 1-Benzyl-2-methyl-5-hydroxymethylimidazole (**3**) was synthesized by a known method (2).

1-Benzyl-2-methylimidazole-5-carboxaldehyde (**6**).

A mixture of 70 g. (0.35 mole) of **3** and 200 g. of active manganese dioxide (5) in 400 ml. of chloroform was stirred under reflux for 4 hours. The residual oil, obtained after filtering of the reaction mixture over Celite and subsequent evaporation of the solvent, was distilled *in vacuo*, yield, 56.2 g. (81%), b.p. 124-128° (0.03); m.p. 51-52° (lit (2) 48-50°).

1-Benzyl-2-methyl-5-(2,2-dimethoxyethyl)imidazole (**7**).

A solution of 40 g. (0.2 mole) of **6** and 21 g. (0.2 mole) of aminoacetaldehyde-dimethylacetal in 200 ml. of toluene was refluxed for 4 hours under continuous removal of water with a Dean-Stark trap. Evaporation of the solvent and subsequent distillation afforded 51.4 g. (90%) of **7**, b.p. 165-170° (0.04); nmr:  $\delta$  2.28 (s, 3, CH<sub>3</sub>), 3.60 (d, 2, =N-CH<sub>2</sub>-), 8.18 (s, 1, -N=CH-); ir (nujol): 1650 cm<sup>-1</sup>  $\nu$  (C=N).

Anal. Calcd. for C<sub>16</sub>H<sub>21</sub>N<sub>3</sub>O<sub>2</sub>: C, 66.88; H, 7.37; N, 14.62. Found: C, 66.75; H, 7.37; N, 14.62.

1-Benzyl-2-methyl-5-(2,2-dimethoxyethylaminomethyl)imidazole (**8**).

To a solution of 72 g. (0.25 mole) of **7** in 700 ml. of methanol was added with stirring at room temperature 12.5 g. (0.63 mole)

of sodium borohydride. The mixture was refluxed for 3 hours. The solvent was then removed, water was added and the product extracted with benzene. Distillation gave 57 g. (79%) of **8**, b.p. 185-195° (0.03); nmr:  $\delta$  2.32 (s, 3, CH<sub>3</sub>), 3.70 (s, 2, -NH-CH<sub>2</sub>-imidazole), 6.96 (s, 1, H-4 imidazole).

Anal. Calcd. for C<sub>18</sub>H<sub>27</sub>N<sub>3</sub>O<sub>2</sub>: C, 66.41; H, 8.01; N, 14.52. Found: C, 66.37; H, 8.06; N, 14.52.

1-Benzyl-2-methyl-5-N-(2,2-dimethoxyethyl)-N-methylamino-methylimidazole (**1b**).

To a solution of 20.2 g. (0.07 mole) of **8** and 7.7 g. (0.08 mole) of triethylamine in 100 ml. of ether was added at 0-5° a solution of 6.6 g. (0.07 mole) of methylchloroformate in 20 ml. of ether. After stirring for 1 hour the precipitate was filtered and the solvent removed. The residue consisted of nearly pure carbamate ester; nmr:  $\delta$  2.30 (s, 3, CH<sub>3</sub>), 3.22 (s, 6, OCH<sub>3</sub>), 3.50 (s, 3, COOCH<sub>3</sub>), 7.00 (s, 1, H-4 imidazole). The crude oil was dissolved in 35 ml. of THF and added dropwise under reflux to a stirred suspension of lithium aluminumhydride (5 g.) in 60 ml. of THF. After 2 hours the mixture was worked up in the usual manner and afforded 10.3 g. (49% of **1b**; b.p. 160-170° (0.02); nmr:  $\delta$  2.20 (s, 3, N-CH<sub>3</sub>), 2.29 (s, 3, CH<sub>3</sub>), 3.22 (s, 6, OCH<sub>3</sub>), 5.31 (s, 2, CH<sub>2</sub>-C<sub>6</sub>H<sub>5</sub>), 6.85 (s, 1, H-4 imidazole).

Anal. Calcd. for C<sub>17</sub>H<sub>25</sub>N<sub>3</sub>O<sub>2</sub>·2HCl: C, 54.26; H, 7.23; N, 11.27. Found: C, 54.35; H, 7.26; N, 11.13.

1-Benzyl-2-methyl-5-(2,2-dimethoxyethylbenzylaminomethyl)imidazole (**1c**).

A solution of 25.3 g. (0.18 mole) of benzoyl chloride in 200 ml. of ether was added dropwise with stirring at 5-10° to a solution of 52 g. (0.18 mole) of **8** in 400 ml. of ether, containing 20.2 g. (0.2 mole) of triethylamine. After stirring for 2 hours at room temperature 630 ml. of 1N sodium hydroxide was added and the

product was extracted with benzene. Washing, drying and evaporation of the solvent afforded the benzoyl compound as an oil; nmr:  $\delta$  2.30 (s, 3,  $CH_3$ ), 3.29 (s, 6,  $OCH_3$ ), 4.77 (s, 2,  $N-CH_2C_6H_5$ ), 7.08 (s, 1, H-4 imidazole); ir (nujol):  $1700\text{ cm}^{-1}$   $\nu$  (C=O). Reduction of the crude oil with lithium aluminumhydride in ether at  $0^\circ$  gave 38.2 g. (56%) of **1c** as an oil; nmr:  $\delta$  2.25 (s, 3,  $CH_3$ ), 3.15 (s, 6,  $OCH_3$ ), 3.53 (s, 2,  $-N-CH_2$ -imidazole), 3.61 (s, 2,  $CH_2-N-CH_2C_6H_5$ ), 6.92 (s, 1, H-4 imidazole).

*Anal.* Calcd. for  $C_{23}H_{29}N_3O_2 \cdot 2HCl$ : C, 61.06; H, 6.91; N, 9.29. Found: C, 60.79; H, 6.93; N, 9.24.

#### 1-Benzyl-2-methyl-5-(2,2-diethoxyethoxymethyl)imidazole (**1a**).

To a solution of 50 g. (0.25 mole) of **3** in 250 ml. of DMF was added 9 g. (ca 0.3 mole) of sodium hydride (80% dispersion in mineral oil). The mixture was stirred and heated at  $60-80^\circ$  until hydrogen evolution had ceased. Then 400 ml. of toluene was added and 65 g. (0.33 mole) of bromoacetaldehyde-diethylacetal and the mixture was refluxed for 24 hours. After that period 2 l. of water was added; the toluene layer was separated and the water phase was extracted twice with ether. The combined organic phases were washed, dried and evaporated. The residual oil was chromatographed (silica gel; chloroform-2% methanol as eluent) and distilled, yield 31.8 g. (40%) of **1a** as an oil; b.p.  $170-175^\circ$  (0.01); nmr:  $\delta$  2.32 (s, 3,  $CH_3$ ), 4.45 (s, 2,  $-O-CH_2-$ imidazole), 5.25 (s, 2,  $-N-CH_2-C_6H_5$ ), 7.02 (s, 1, H-4 imidazole).

*Anal.* Calcd. for  $C_{16}H_{26}N_2O_3$ : C, 67.90; H, 8.23; N, 8.80. Found: C, 67.63; H, 8.06; N, 8.60.

#### 1-Benzyl-2-methyl-5-chloromethylimidazole Hydrochloride (**4**).

A solution of 13.1 g. (0.11 mole) of thionyl chloride was added with stirring at  $10^\circ$  to a solution of 20.2 g. of **3** in 100 ml. of chloroform. After one hour the solvent was removed and the residue was treated with 150 ml. of ether, thus affording 24 g. (93%) of **4**, m.p. (2-propanol)  $182-183^\circ$ .

*Anal.* Calcd. for  $C_{12}H_{13}ClN_2 \cdot HCl$ : C, 56.01; H, 5.48; N, 10.89. Found: C, 56.28; H, 5.66; N, 10.87.

#### 1-Benzyl-2-methyl-5-mercaptomethylimidazole (**5**).

A mixture of 25.7 g. (0.1 mole) of **4** and 7.6 g. (0.1 mole) of thiourea in 150 ml. of ethanol was refluxed for 6 hours. Evaporation of the solvent left a viscous oil, which upon treatment with diisopropylether afforded 28.1 g. (85%) of essentially pure thiouronium salt, m.p.  $198-202^\circ$ .

*Anal.* Calcd. for  $C_{13}H_{16}N_4S \cdot 2HCl$ : C, 46.85; H, 5.41; N, 16.82. Found: C, 47.01; H, 5.42; N, 16.81.

The thiouronium salt was dissolved in 350 ml. of 5*N* sodium hydroxide and refluxed for 1 hour. The reaction mixture was then neutralized with acetic acid and extracted with chloroform. After washing, drying and evaporation of the solvent 16.4 g. (74%) of **5** left as an oil which rapidly decomposed on standing. The crude product was pure enough for the next step (more than 90% based upon nmr); nmr:  $\delta$  2.38 (s, 3,  $CH_3$ ), 3.58 (s, 2,  $-S-CH_2-$ imidazole), 6.90 (s, 1-H-4 imidazole).

#### 1-Benzyl-2-methyl-5-(2,2-diethoxyethylmercaptomethyl)imidazole (**1d**).

To a suspension of 2.4 g. (0.08 mole) of sodium hydride (80% dispersion in mineral oil) in 50 ml. of DMF was added with stirring a solution of 16.5 g. (0.076 mole) of **5** in 30 ml. of DMF. After hydrogen evolution had ceased stirring was prolonged for 1 hour. Then a solution of 16.7 g. (0.085 mole) of 2-bromoacetaldehyde-diethylacetal in 20 ml. of DMF was added dropwise at  $30^\circ$ . After stirring for additional 5 hours water (500 ml.) was added and the product was extracted with ether. Chromatography and subsequent

distillation yielded 12.1 g. (48%) of **1d** as a viscous oil, b.p.  $185-195^\circ$  (0.02); nmr:  $\delta$  1.20 (t, 6,  $-OCH_2-CH_3$ ), 2.33 (s, 3,  $CH_3$ ), 3.70 (s, 2,  $-S-CH_2$ -imidazole).

*Anal.* Calcd. for  $C_{18}H_{26}N_2O_2S$ : C, 64.67; H, 7.78; N, 8.38. Found: C, 64.88; H, 7.77; N, 8.57.

#### 1-Benzyl-2-methyl-4-hydroxy-6-thia-4,5,6,7-tetrahydrobenzimidazole (**2d**).

A mixture of 3.34 g. (0.01 mole) of **1d**, 12 g. of sodium acetate, 10 ml. of acetic acid and 40 ml. of water was refluxed for 2 hours. The solution was rendered basic with 3*N* sodium hydroxide and extracted with chloroform. Chromatographing the product (silica gel; chloroform-10% methanol as eluent, Rf 0.24) afforded 1.69 g. (65%) of **2d**, m.p. (toluene)  $161-162^\circ$ ; nmr:  $\delta$  2.39 (s, 3,  $CH_3$ ), 3.00 (m, 2,  $-S-CH_2-CHOH-$ ), 3.40 (s, 2,  $-S-CH_2-$ imidazole), 5.45 (s, 1,  $OH$ ); ir (potassium bromide):  $3120\text{ cm}^{-1}$   $\nu$  ( $OH$ ).

*Anal.* Calcd. for  $C_{14}H_{16}N_2OS$ : C, 64.58; H, 6.19; N, 10.76. Found: C, 64.31; H, 6.11; N, 10.50.

#### 1,6-Dibenzyl-2-methyl-4-hydroxy-6-aza-4,5,6,7-tetrahydrobenzimidazole (**2c**).

A mixture of 3.79 g. (0.01 mole) of **1c**, 12 g. of sodium acetate, 10 ml. of acetic acid and 40 ml. of water was refluxed for 8 hours. The solution was neutralized with 3*N* sodium hydroxide and extracted with chloroform. After washing, drying and stripping off the solvent, the residual oil was chromatographed (silica gel; chloroform-10% methanol as eluent, Rf 0.15) and afforded 0.70 g. (21%) of **2c** as an oil; nmr:  $\delta$  2.33 (s, 3,  $CH_3$ ), 2.92 (d, 2,  $-N-CH_2-CHOH-$ ), 3.34 (d, 2,  $-N-CH_2$ -imidazole), 3.71 (s, 2,  $-N-CH_2-C_6H_5$ ); m.p. (bis HCl-salt)  $228-230^\circ$  (2-propanol).

*Anal.* Calcd. for  $C_{21}H_{23}N_3O \cdot 2HCl$ : C, 62.07; H, 6.20; N, 10.34. Found: C, 61.60; H, 6.38; N, 10.77.

#### 1-Benzyl-2,6-dimethyl-4-hydroxy-6-aza-4,5,6,7-tetrahydrobenzimidazole (**2b**).

A solution of 3.03 g. of **1b** (0.01 mole), 12 g. of sodium acetate and 10 ml. of acetic acid in 40 ml. of water was refluxed for 8 hours. The mixture was made basic with 3*N* sodium hydroxide and extracted with chloroform. The product obtained after washing, drying and stripping off the solvent, was chromatographed over silica gel (chloroform-10% methanol as eluent, Rf 0.11) and gave 0.59 g. (32%) of pure **2b**, m.p. (dioxane-diisopropylether)  $152-153^\circ$ ; nmr:  $\delta$  2.32 (s, 3,  $CH_3$ ), 2.45 (s, 3,  $CH_3$ ), 2.80 (m, 2,  $NCH_2-CHOH$ ), 3.28 (d, 2,  $-N-CH_2$ -imidazole), 4.30 (s, 1,  $OH$ ), 4.80 (t, 1,  $CHOH$ ).

*Anal.* Calcd. for  $C_{15}H_{19}N_3O$ : C, 70.01; H, 7.44; N, 16.33. Found: C, 69.90; H, 7.44; N, 16.27.

#### 1-Benzyl-2-methyl-4-hydroxy-6-oxa-4,5,6,7-tetrahydrobenzimidazole (**2a**).

A solution of 2.94 g. (0.01 mole) of **1a**, 12 g. of sodium acetate and 10 ml. of acetic acid in 40 ml. of water was refluxed for 8 hours. After cooling, the reaction mixture was brought to pH 8 with 3*N* sodium hydroxide and extracted with chloroform. Washing, drying and evaporation of the solvent left an oil which upon chromatographing over silica gel (chloroform-10% methanol as eluent, Rf 0.20) yielded 0.68 g. (28%) of **2a**, m.p. (toluene)  $153-154^\circ$ ; nmr:  $\delta$  2.38 (s, 3,  $CH_3$ ), 3.95 (q, 2,  $-O-CH_2-CHOH$ ), 4.48 (d, 2,  $-O-CH_2$ -imidazole), 4.78 (t, 1,  $CHOH$ ), 4.93 (s, 2,  $-N-CH_2-C_6H_5$ ), 5.32 (s, 1,  $OH$ ).

*Anal.* Calcd. for  $C_{14}H_{16}N_2O_2$ : C, 68.83; H, 6.60; N, 11.47. Found: C, 68.76; H, 6.54; N, 11.42.

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